



# 1 Design methodology to determine water quality monitoring strategy of 2 surface water treatment plants

3  
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## 8 9 Abstract

10 Primary goal of a drinking water company is to produce safe drinking water fulfilling the quality  
11 standards defined by national and international guidelines. To ensure the produced drinking water meets  
12 the quality standards, sampling of the drinking water is carried out on a regular (almost daily) basis. It  
13 is the dilemma that the operator wishes to have a high probability of detecting a bias while minimizing  
14 his measuring effort. In this paper a seven step design methodology is described on how to come to an  
15 optimised water quality monitoring scheme. It was shown that the previous on-line monitoring program  
16 of a WTP could be optimised. Besides using soft-sensors as surrogate sensors for parameters currently  
17 not available on-line, they can also provide a cost effective alternative when used to determine multiple  
18 parameters required through one single instrument.

## 19 20 Keywords

21 Data requirements; design methodology; model-based optimization; soft-sensors

## 22 23 INTRODUCTION

24 Primary goal of a drinking water company is to produce safe drinking water fulfilling the quality standards defined  
25 by national and international guidelines. To ensure the produced drinking water meets the quality standards,  
26 sampling of the drinking water is carried out on a regular (almost daily) basis.

27  
28 Common practice is that (drinking) water treatment plants (WTPs) are designed in such a robust way that the  
29 effluent quality can be guaranteed without direct control on the incoming water quality (Vanrolleghem and Lee,  
30 2003;Bosklopper et al., 2004). A WTP consists of several individual treatment steps placed in series, with every  
31 treatment step being responsible for the removal (or addition) of certain compounds. All the interactions between  
32 the processes ask for an integrated plant-wide approach, optimizing the effluent quality and operational costs  
33 (Bosklopper et al., 2004;Nopens et al., 2010).

34  
35 Van der Helm et al. (2008b) investigated three possible objectives for plant-wide optimization of operation of  
36 existing WTPs and concluded that the objective for integrated optimization should be the improvement of water  
37 quality and not a reduction in environmental impact and costs. The effects of these latter two are negligible  
38 compared to the environmental impact and costs for the society as a whole when more bottled water is used for  
39 drinking water as a result of insufficient (confidence in) tap water quality.

40  
41 Direct control of water quality becomes more and more important as a result of more stringent criteria and the  
42 deterioration of source water (Vanrolleghem and Lee, 2003;van Schagen et al., 2010). Especially WTPs that use  
43 surface water as a source, experience increased pollution in the form of organic micropollutants and increased  
44 organic matter concentrations present in the surface water bodies (Verliefde et al., 2007;Bertelkamp et al., 2014).  
45 Besides, large fluctuations in water temperature and water quality can be noticed, which increases the need for  
46 direct control of the WTP.

47  
48 Nowadays, many WTPs are monitored and controlled by SCADA (Supervisory Control and Data Acquisition)  
49 systems (Jansen et al., 1997). The functions of SCADA systems for WTPs include: (1) collection of on-line  
50 measurement data, (2) surveillance of the measuring chain including operations and (3) process control and other  
51 relevant operations (Gunatilaka and Dreher, 2003). On-line measurements are the first indicators that give the



1 operators information about the state the plant is in. Besides on-line measurements, laboratory measurements are  
2 taken at a regular interval to check that the produced drinking water meets the quality standards set by national  
3 and international guidelines. However, the time between sampling and results takes at least one day. This delay in  
4 results and interval between measurements makes it difficult to use the laboratory measurements for real-time  
5 control of a treatment plant (van de Ven et al., 2010). In addition, it should not be underestimated that erroneous  
6 control and measurement devices can also cause disturbances (van Schagen et al., 2010).

7  
8 Retrieving reliable and robust on-line information is therefore important in order to be able to control a WTP. This  
9 information can be retrieved from on-line sensors that measure a specific parameter directly, but also from generic  
10 sensors that give indirect information. Roccaro et al. (2008), Rieger et al. (2004) and van den Broeke et al. (2008)  
11 showed the ability of UV-Vis spectra measurements, measuring the absorbance of ultraviolet or visible light, to  
12 estimate different parameters such as chlorine decay, nitrite and nitrate, ozone and assimilable organic carbon  
13 (AOC) concentrations. These estimations were derived from algorithms developed, based on a change in UV-Vis  
14 absorbance during a treatment step and laboratory measurements, using principal component analysis followed by  
15 partial least squares regression. These types of generic sensors are so-called soft-sensors, sensors that require  
16 software to give the required information. Juntunen et al. (2013) developed a soft-sensor to predict the turbidity in  
17 treated water and to find the most significant variables affecting turbidity.

18  
19 A soft-sensor can be developed in different ways, based on black box, grey box or white box modeling. The black  
20 box approach is characterized by an empirical relation between the input and output. The relations are derived  
21 from historical, full-scale plant, data. Thus, such a soft-sensor can only be applied in the situation where it has  
22 been developed for, since a black box model is not valid when a process is operated outside the boundaries of  
23 calibration (Kano and Nakagawa, 2008). Because the operation of a WTP is relatively constant, the calibration  
24 dataset is normally rather limited, hampering the application of black box modeling. Grey box models are a  
25 combination of black box models and white box models, such that it contains some more insight into the system  
26 through the white box model, while still some parts of the model are data driven (Zyngier et al., 2001). White box  
27 models mathematically describe the physical-chemical processes that take place in the treatment process.  
28 Developing these models is time consuming, however, when developed, the process knowledge on the processes  
29 are captured, leading to more generically applicable models (van der Helm and Rietveld, 2002).

30  
31 Optimized control can only be reached if there is a high probability of detecting a bias in the operation of the WTP.  
32 At the same time, from an economical perspective, the data should be obtained with minimal measuring efforts  
33 and costs. Understanding the requirements with respect to on-line monitoring and data reliability is a first step  
34 towards direct control of the drinking water production based on the incoming water quality. Therefore, in this  
35 paper a design methodology is described on how to come to an optimized water quality monitoring scheme to  
36 support direct control. This will be explained by means of a case study for a WTP.

## 37 38 **MATERIALS AND METHODS**

### 39 40 **Design methodology**

41 Van Schagen et al. (2010) developed a methodology for the design of a control system for drinking water treatment  
42 plants. This methodology was based on experiences with control design procedures for chemical plants and was  
43 modified to fit the main objectives of a drinking water treatment plant. In the basis, the same methodology was  
44 used for the design of an optimized water quality monitoring scheme. The methodology takes into consideration  
45 1) the objectives, 2) operational constraints and 3) disturbances. These first three steps determine the required  
46 water quality parameters. The subsequent steps help to determine the conditions the water quality information  
47 should comply with:

- 48
- 49 1. Determine treatment step objectives;
- 50 2. Determine operational control options;
- 51 3. Determine water quality parameters taking into consideration both process and control aspects;



- 1 4. Identify process characteristics;
- 2 5. Evaluate available (indirect) measurements;
- 3 6. Determine individual monitoring strategy per treatment step.
- 4 7. Determine integrated monitoring strategy of treatment plant.

5  
6 ***Treatment step objectives***

7 The treatment step objectives depend on the feed water quality and the type of treatment step considered. The  
8 overall objective of a drinking water treatment plant is the production of safe drinking water fulfilling the quality  
9 standards defined by national and international guidelines. The main objective of a treatment step for an existing  
10 plant should be the focus on water quality and less on the chemical or energy consumption (van der Helm et al.,  
11 2008b). Therefore it should be evaluated which parameters, present in the feed water quality, can be influenced  
12 per treatment step. In order to do so process knowledge on the different treatment steps is indispensable (Poch et  
13 al., 2004). Van Schagen (2009) indicated that mathematical models are a powerful tool to evaluate the sensitivity  
14 to process objectives and disturbances and help find the appropriate controlled variables.

15  
16 ***Operational control options***

17 Depending on the design of the treatment step certain operational control options are available to make changes to  
18 the treatment process. Examples of operational control options are the change in chemical dosage, flow division  
19 and backwash and regeneration frequency. The primary focus is on the operational changes that can be performed  
20 within the existing plant lay-out.

21  
22 ***Required water quality parameters***

23 Based on the treatment step objectives and existing operational control options, the water quality parameters that  
24 are influenced by the treatment step are determined. Ideally these water quality parameters should be monitored.  
25 Besides the water quality parameters that are influenced by a treatment step, there are water quality parameters  
26 that influence the efficiency of a treatment step. For example, the water temperature has an effect on the ozone  
27 decay rate. The decay rate increases with increasing temperatures (Elovitz et al., 2000). This may result in a higher  
28 required ozone dose in summer time, taking into consideration that the disinfection requirements are also different  
29 with different temperatures.

30  
31 ***Process characteristics***

32 The required monitoring frequency and sensitivity of the selected water quality parameters may also vary  
33 depending on the process characteristics. The process characteristics describe the time interval during which  
34 changes occur and the order of magnitude in which changes occur. For instance, the contact time in an ozone  
35 reactor can vary from a couple of minutes to one hour, depending on the dimensions, while the time between two  
36 regeneration cycles of activated carbon typically is expressed in years. These different reaction times require  
37 different measurement frequencies. The order of magnitude relates to the required accuracy of the measurement.  
38 For example, ozone typically degrades quickly in water due to the reaction with organic compounds in the water.  
39 This determines that the required measurement sensitivity and accuracy should be high.

40  
41 ***Evaluate available measurements for the identified water quality parameters***

42 Based on the evaluation of the required water quality parameters and existing process characteristics the available  
43 (on-line) measurements should be evaluated. A wide range of measurements exist for determining water quality  
44 parameters, from certified laboratory measurements to on-line measurements. Depending on the variability of the  
45 process, the turnaround time of laboratory measurements is not always fast enough. To come to an optimal water  
46 quality monitoring scheme also on-line water quality sensors should be considered. In this study the following  
47 evaluation criteria for the available on-line sensors were assessed:

- 48 Easiness; is the sensor easy to use, is the measuring principle easy to understand;
- 49 Sensitivity; is the measurement range sensitive enough;
- 50 Maintenance; does the sensor require much maintenance;



1 Costs for laboratory measurements as well as the purchasing and maintenance costs for on-line sensors were  
2 indicated. Besides on-line sensors developed to measure one specific parameter, available surrogate sensors, used  
3 to estimate a water quality parameter value, and soft-sensors were assessed.

4  
5 **Determine individual monitoring strategy per treatment step**

6 The individual monitoring strategy defines which water quality parameters per treatment step should be monitored,  
7 with a selected frequency and location. The evaluation, of available measurements for the identified water quality  
8 parameters forms the basis for the monitoring strategy, subsequently ranked by the most critical parameters in the  
9 treatment plant. Criticality is determined by two factors, 1) parameters of which the measured concentrations are  
10 close to the not to exceed limit and 2) parameters that can be potentially harmful to human health.

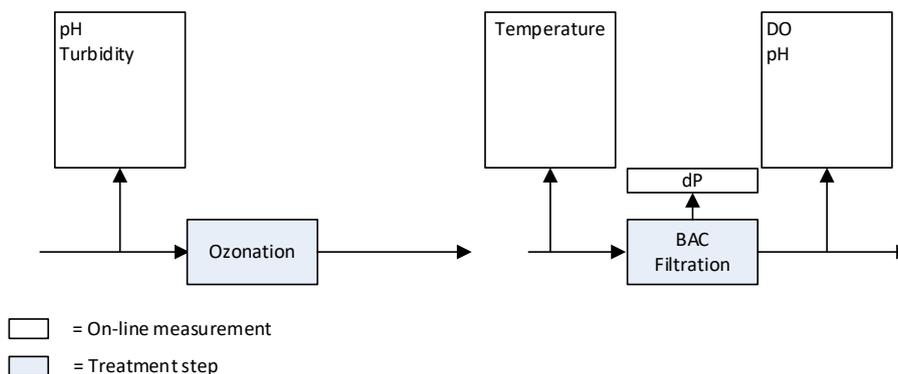
11  
12 **Determine integrated monitoring strategy of treatment plant**

13 The integrated monitoring strategy defines which water quality parameters are monitored, taking into  
14 consideration the interaction between the different individual treatment processes. The evaluation, of available  
15 measurements for the identified water quality parameters forms the basis for the monitoring strategy, again ranked  
16 by the most critical parameters in the treatment plant. The monitoring strategy can be imbedded into the process  
17 control strategy to ensure optimized control based on the most critical parameters.

18  
19 **Case study: Ozonation and biological activated carbon filtration at Waternet**

20 At the production location Weesperkarspel of Waternet, the water cycle company of Amsterdam and surroundings,  
21 ozonation, pellet softening, biological activated carbon (BAC) filtration and slow sand filtration are the main steps  
22 in the production of safe drinking water. The feed water is humics rich seepage water from the Bethune polder,  
23 which is pre-treated by coagulation, sedimentation, approximately 100 days retention in a lake reservoir followed  
24 by rapid sand filtration, before it is transported to the Weesperkarspel treatment plant. At Weesperkarspel, the  
25 production of drinking water is roughly divided into two parallel lanes, each consisting of several individual  
26 reactors/filters per treatment step. The control actions can be modified at individual level, however, for the purpose  
27 of this Chapter it has been chosen to focus on the mixed influent and effluent only and not on the individual  
28 reactor/filter level. The treatment processes ozonation and BAC filtration have been evaluated. These processes  
29 are frequently applied at surface WTPs and are susceptible to changes in the feed water quality. Besides, these  
30 processes have several control options and an interaction between the two processes exists.

31  
32 Previously, the following on-line measurements were installed to monitor the ozonation and BAC filtration process  
33 (Figure 1).  
34



35  
36 **Figure 1 Previous installed on-line measurements ozonation and BAC filtration at Weesperkarspel treatment plant**  
37



1 IpH and turbidity were monitored at the influent of the ozonation step. The temperature was monitored in the  
2 influent of the BAC filtration. After BAC filtration dissolved oxygen (DO) and pH were measured, and the  
3 pressure drop was recorded over each of the individual BAC filters.

## 4 5 **RESULTS**

6 The results of the evaluation of each step, to come to an optimised water quality monitoring scheme, are described  
7 below, followed by a discussion on the outcomes of the assessment versus the previous and current monitoring  
8 strategy. Research carried out at the pilot plant of Weesperkarspel was used to obtain full understanding of the  
9 processes taking place and enabling the determination of the objectives and required water quality parameters.

### 10 11 *Treatment plant objectives*

12 In general the primary objective of ozonation is disinfection (von Gunten, 2003b). Besides, ozonation is frequently  
13 used for the oxidation of organic micro pollutants, taste, odour and colour producing products and natural organic  
14 matter (NOM), transforming higher molecular weight compounds into lower molecular weight compounds. For  
15 the ozonation step at Weesperkarspel, the specific objectives are disinfection and oxidation of NOM (van der Helm,  
16 2007).

17  
18 The general objective of activated carbon is the removal of organic micropollutants, removal of precursors of  
19 disinfection by-products and the removal of organic compounds causing colour, taste and odour issues (van der  
20 Aa et al., 2011). When activated carbon is preceded by a pre-oxidation step, the biological activity in the water  
21 and on the activated carbon is enhanced, resulting in BAC filtration. At the same time ozonation increases the  
22 polarity, resulting in a decrease in adsorption capacity (Sontheimer et al., 1988). As a result, NOM is removed  
23 through both biodegradation and adsorption. At Weesperkarspel the purpose of BAC filtration is the removal of  
24 organic matter, to prevent biological growth in the distribution system and to remove toxicity, taste and odour  
25 causing compounds (Graveland, 1996). Besides, the BAC filters remove the carry-over from the preceding pellet  
26 softening step.

### 27 28 *Operational control options*

29 The production flow is controlled by the demand for drinking water. The buffering capacity in the treatment plant  
30 is the clean water storage reservoirs situated before the water is distributed to the customers (van Schagen et al.,  
31 2010). To ensure sufficient reliability, the treatment plant is set up in a redundant way with multiple lanes operated  
32 in parallel. It is possible to change the flow division over the different production lanes, however this is only done  
33 when one of the lanes has less treatment capacity or is out of production due to e.g. maintenance. Therefore, in  
34 this case, production flow was not considered as a control action.

35  
36 The only remaining control action for ozonation is the ozone dosage. The ozone dosage is obtained by a  
37 combination of ozone in gas concentration and the gas flow. Both parameters can be adjusted to obtain the desired  
38 ozone dosage.

39  
40 For BAC filtration the control actions within the existing treatment setup are the backwash frequency, currently  
41 operated at every couple of days till once a month interval per filter and backwash program, currently a  
42 combination of air and water is used. The activated carbon is regenerated every year to three years. Carbon dioxide  
43 is dosed before the BAC filters to correct for any high pH resulting from the caustic soda dosage in the pellet  
44 softeners. This control action is thus more related to the functioning of the pellet softeners and therefore not  
45 included in the overview provided in Figure 1. A high pH could negatively affect the biodegradation efficiency  
46 (Seredyńska-Sobecka et al., 2006) and promotes precipitation of calcium carbonate on the activated carbon grains.  
47 Oxygen and caustic soda can be dosed in the effluent of the BAC filters to correct low pH and oxygen  
48 concentrations as a result of the biological activity in the filters.

### 49 50 *Required water quality parameters*



1 As indicated previously, ozone is an unstable oxidant in water. Ozone decomposition in water consists of a fast  
2 initial phase (seconds range) and second phase (minutes range) during which ozone concentration decreases via  
3 first order kinetics and disinfection of the more resistant pathogenic microorganisms takes place (von Gunten,  
4 2003a;van der Helm et al., 2008a). A commonly used method to determine the disinfection capacity of ozonation  
5 is by calculating the exposure of pathogens to ozone, expressed as the Ct value, a product of the (residual)  
6 concentration of the disinfectant (C), in this case ozone and contact time (t) (WHO, 2008).

7  
8 Water quality parameters that influence the efficiency of the ozonation step are temperature, pH and, for  
9 Weesperkarspel relevant, scavengers such as NOM concentration and character (von Gunten, 2003a). A  
10 measurement commonly used to indicate the NOM concentration is the dissolved organic carbon (DOC)  
11 concentration. The DOC concentration is determined by filtering the sample over a 0.45 µm filter and measuring  
12 the total organic carbon (TOC) concentration. In order to assess the character of NOM, the specific UV absorbance  
13 (SUVA) can be calculated by dividing the UV absorbance measured at a wavelength of 254 nm (UV254) by the  
14 DOC concentration (van der Helm et al., 2008b;Edzwald and Tobiasson, 1999). These water quality parameters  
15 play a role in the ozone dosage required to achieve the desired disinfection and should therefore be monitored. For  
16 Weesperkarspel it was determined that disinfection of Giardia, Cryptosporidium and Campylobacter is sufficient  
17 to determine the microbiological safety of the water (van der Helm et al., 2008b). To be able to monitor the  
18 efficiency of the ozonation step, at least one of the following parameters should be measured:

- 19 ▪ Pathogenic mirco-organisms such as Cryptosporidium, Giardia and Campylobacter.
- 20 ▪ Ozone concentration at different contact times, to be able to determine the Ct value (van der Helm et al., 2009);

21  
22 During ozonation disinfection by-products are formed. The oxidation of NOM promotes the presence of AOC  
23 concentration in water (van der Kooij et al., 1989). AOC promotes regrowth of bacteria in a distribution system,  
24 amongst others, and, therefore, should be sufficiently removed in the subsequent treatment steps. Water without  
25 residual chlorine is considered to be biologically stable if the AOC concentration is below 10 µg Acetate-C/L,  
26 whereas water with residual chlorine is defined as biologically stable for AOC concentrations below 50 µg Acetate-  
27 C/L (van der Kooij, 1992;Escobar et al., 2001). Besides AOC, bromate is formed if bromide is present in the feed  
28 water. Bromate is possibly (IARC, 1999) or probable (USEPA, 2018) carcinogenic to humans.

29  
30 During BAC filtration, biodegradation takes place by microorganisms, present on the external surface and in the  
31 macro-pores of the BAC filter grains, that biodegrade the NOM in the water (Servais et al., 1994). The activity of  
32 the microorganisms (biomass) determines the degradation rate of NOM (Lazarova and Manem, 1995). The activity  
33 and concentration of the biomass depends on the concentration of nutrients (carbon, phosphate and nitrogen), the  
34 dissolved oxygen concentration, temperature, pH and residual disinfectant in the feed water (Simpson, 2008). Uhl  
35 and Gimbel (2000) described that for the biological removal of ammonia, the deposit of bacterial cells from the  
36 influent was necessary to maintain a solid biofilm. However for Weesperkarspel it was shown that the feed in  
37 bacterial cells to the BAC filters was not necessary to obtain a sufficient biodegradation efficiency (Ross et al.,  
38 2019), hence no on-line measurement of ATP or flowcytometry was required. Besides biodegradation taking place,  
39 adsorption of NOM and toxic, colour, taste and odour compounds takes place. In addition, at Weesperkarspel,  
40 BAC filtration is simultaneously applied for the removal of suspended solids and carry-over. Due to clogging of  
41 the filter bed by suspended solids, carry-over and in some cases biomass, the filters need to be backwashed  
42 frequently. The pressure drop over the filters and turbidity in the effluent indicates the state the filter is in, and  
43 whether it needs to be backwashed. In case of Weesperkarspel the pressure drop is the determining parameter.

#### 44 **Process characteristics**

45  
46 Ozone is dosed to the water, after which reaction takes place in the seconds to minutes range. A change in ozone  
47 dose or change in feed water quality can have an immediate effect on the effluent quality. In the past, the dosing  
48 strategy was determined by the water temperature, with two different set points, below 12 °C and above 12 °C.  
49 Van der Helm et al. (2009) suggested that this negatively influenced the disinfection during ozonation. However,  
50 more detailed research by Wiersema (2018) could not confirm this. Since ozonation is one of the main processes  
51 that can achieve disinfection, high frequency monitoring is required enabling direct control of the ozonation step.



1  
2 In contrast to ozonation, BAC filtration is not a dosing process, but a separation/degradation process by means of  
3 filtration, adsorption and biodegradation. The different processes all have their associated time intervals. The  
4 shortest time interval is the clogging of the filters, which, depending on the location in the treatment train, needs  
5 to be carried out every couple of days till once a month. Backwashing occurs based on pressure drop over the filter  
6 or after a maximum period of time. The pressure drop should be monitored on a regular basis.

7  
8 As indicated in the required water quality parameters section, the activity of the biomass present on the carbon  
9 grains determines the biodegradation efficiency. Ross et al. (2019) showed that a change in feed water quality does  
10 not necessarily result in a change in effluent quality, hence there is no direct need for close monitoring of the filters.  
11 In case the feed water quality changes for a longer period of time, the biomass will adopt itself to the new situation,  
12 which can take up to 2-3 months (Servais et al., 1994).

13  
14 Depending on the NOM loading, the activated carbon starts showing break-through of organic micro pollutants  
15 and pesticides after a run time of 6-9 months if no biodegradation takes place, while if biodegradation takes place  
16 this can last up to 2-5 years before the activated carbon needs to be regenerated (Simpson, 2008). Although BAC  
17 filters have proven their ability to intercept sudden changes in water quality, the DO can be used as an indicator  
18 for the biological activity in the filter and identifying any disruptions taking place (van Schagen, 2009).

#### 19 20 ***Evaluate available measurements for the identified water quality parameters***

21 A summary of the required water quality parameters, as determined in the paragraphs describing the water quality  
22 parameters, can be found in the first column of Table 1 (ozonation) and Table 2 (BAC filtration). In the second  
23 column it is indicated per parameter if an on-line measurement, able to measure at the limit of detection required,  
24 is available. Depending on the monitoring frequency required, as described in the process characteristics  
25 paragraphs, it was determined if a parameter should be available on-line. If the monitoring frequency should be  
26 daily or more, it was indicated with a yes in the third column. To gain a better understanding of the applicability  
27 of the on-line sensors, the ease of use, sensitivity and maintenance requirements were evaluated in columns four  
28 through six. The costs related to a measurement in lab and installation of an on-line sensor are listed in column  
29 seven.

30  
31 Evaluation of availability on-line sensors and its characteristics was based on literature research, indicated by the  
32 references included per parameter. Besides on-line sensors that measure one specific parameter, available related  
33 surrogate parameters (column eight) and soft-sensors (column nine) were also captured. It should be noted that for  
34 some surrogate parameters and soft-sensors a start concentration is required first before the concentration of the  
35 requested parameter can be estimated.

36



**Table 1 Summary water quality parameters required to monitor ozonation and associated available on-line sensors**

Parameter	On-line available	On-line required	Easy	Sensitive enough	Maintenance	Costs lab/online	Surrogate parameters	Soft-sensor available
pH	Yes (Banna et al., 2014)	Yes	Yes	Yes	Moderate, needs regular calibration	lab/online: low	No	Yes through water quality (WQ) modeling after dosages of a base or acid based on measured influent pH (van Schagen et al., 2009)
Temperature	Yes (Banna et al., 2014)	Yes	Yes	Yes	Low	lab/online: low	No	No
DOC	Yes via TOC measurement (Hall et al. 2007)	Yes	Moderate	Yes	High, 0.45 µm filters and reagents are required to be replaced	lab: moderate online: high	UV <sub>254</sub> or a UV <sub>280</sub> , UV wavelength at 254 or 280 nm related to reactivity of the organic carbon with ozone (Westerhoff et al., 1999)	Yes, based on range of UV wavelengths (Langergraber et al., 2003)
UV <sub>254</sub>	Yes (Hach, 2018)	Yes	Yes	Yes	Yes	lab: low online: moderate	UV/Vis measurement, measuring all wavelengths between 200 – 735 nm	n.r.
Pathogenic micro-organisms	No	Yes	n.a.	n.a.	n.a.	lab: high online: n.a.	Ct value related to inactivation of Giardia after measuring influent concentration (USEPA, 1989)	Yes, Ct value estimation by means of WQ modeling (van der Helm et al., 2009) or algorithm based UV/Vis-spectra measurements after measuring influent concentration (Ross et al., 2016)
AOC	No	Yes	n.a.	n.a.	n.a.	lab: high online: n.a.	Yes (Hammes and Egli, 2005)	Yes, through WQ modeling by van der Helm et al. (2009) or algorithm based on UV/Vis-spectra measurements (Ross et al., 2016)
Bromate	No (ThermoFisher, 2018)	Yes	n.a.	n.a.	n.a.	lab: moderate online: n.a.	Yes, Ct value has linear relationship with bromate (van der Helm et al., 2008a)	Yes, through WQ modeling by van der Helm et al. (2009) or UV/Vis-spectra measurements (Ross et al., 2016)
Bromide	No	No	n.a.	n.a.	n.a.	lab: moderate online: n.a.	n.r.	n.r.
Ozone concentration in water	Yes (Hach, 2018)	Yes	Moderate	No	Moderate, regular cleaning required	lab/online: moderate	Yes, UV absorbance from 185-350 nm (Molina and Molina, 1986)	Yes, developed based on UV measurement (van den Broeke et al., 2008)

n.a.= not applicable, n.r. = not required.



**Table 2 Summary water quality parameters required to monitor BAC filtration and associated available on-line sensors**

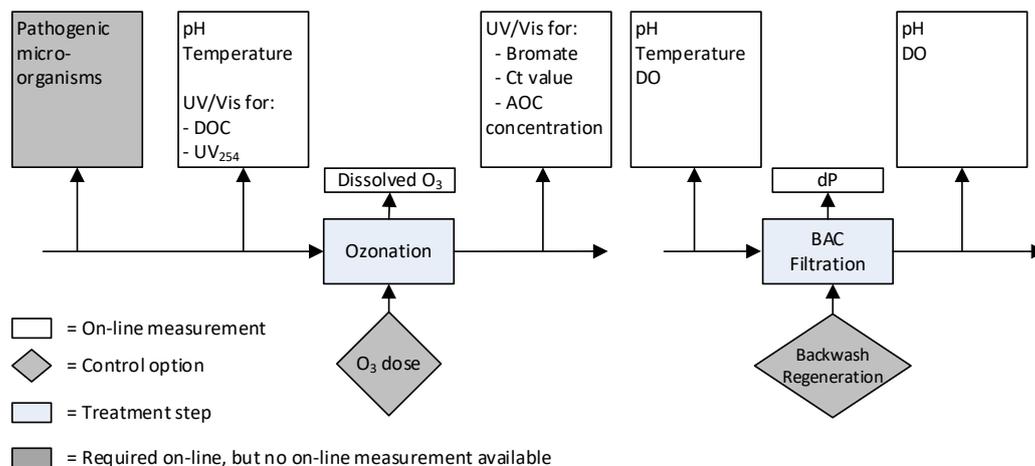
Parameter	On-line available	On-line required	Easy	Sensitive enough	Maintenance	Costs lab/online	Surrogate parameters	Soft-sensor available
DO	Yes (Banna et al., 2014)	Yes	Yes	Yes	Low	lab/online: low	No	No
Phosphate	Yes (Schlegel and Baumann, 1996;Hach, 2018)	No	Yes	No	Moderate, reagents are required to be replaced	lab: moderate online: moderate	n.r.	n.r.
Nitrogen	No	No	n.a	n.a.	n.a	lab: moderate online: n.a.	n.r.	n.r.
DOC	Yes via TOC measurement (Hall et al. 2007)	No	Moderate	Yes	High, 0.45 µm filters and reagents are required to be replaced	lab: moderate online: high	n.r.	n.r.
AOC	No	No	n.a.	n.a.	n.a.	lab: high online: n.a.	n.r.	n.r.
Viable bacterial cells	Yes (Besmer et al., 2017)	No	Moderate	Yes	Moderate	lab: moderate online: high	n.r.	n.r.
pH	Yes (Banna et al., 2014)	Yes	Yes	Yes	Moderate, needs regular calibration	lab/online: low	No	Yes through water quality (WQ) modeling after dosages of a base or acid based on measured influent pH (van Schagen et al., 2009)
Temperature	Yes (Banna et al., 2014)	Yes	Yes	Yes	Low	lab/online: low	No	No
Pressure drop	Yes (van Schagen et al., 2008)	Yes	Yes	Yes	Low	lab: moderate online: low	n.r.	n.r.

n.a.= not applicable, n.r. = not required.



1 **Determination of individual monitoring strategy per treatment step**

2 Figure 2 shows the individual monitoring strategy per treatment step determined by the water quality assessment  
 3 captured in Table 1 for ozonation and Table 2 for BAC filtration. The results are described in detail below.  
 4



5  
 6 **Figure 2 Required on-line water quality information for optimized monitoring and control of ozonation and BAC**  
 7 **filtration**  
 8  
 9

10 **pH, temperature, and DO**

11 Being compliance parameters published by the WHO, there are sufficient on-line sensors available to measure the  
 12 pH, temperature, and DO. These sensors are relatively easy to use and sensitive enough. The pH sensor requires  
 13 frequent maintenance. The costs of measurement, either on-line or in laboratory are low. The efficiency of ozone  
 14 is, amongst others, determined by the pH and temperature and should therefore be monitored continuously. The  
 15 DO and pH are a continuously controlled effluent parameter in BAC filtration. The pressure drop indicates if a  
 16 filter needs to be backwashed. The DO and pH are an indicator for the biological activity in the filter and capable  
 17 of identifying any disruptions taking place (van Schagen, 2009).

18 **DOC and UV<sub>254</sub>**

19 The NOM concentration, measured through DOC, is a scavenger and does directly interfere with disinfection,  
 20 requiring to be monitored in the influent of the ozone step. The used ozone dosages hardly affect the DOC  
 21 concentration, limiting the need for monitoring downstream of the ozone step (van der Helm et al., 2008a). For  
 22 TOC there is an on-line sensor available which measures sensitive enough. By inclusion of a 0.45 µm filtration  
 23 step the DOC is determined. It does require frequent maintenance for replacing the 0.45 µm filters and reagents.  
 24 The on-line sensors are still expensive whilst the lab measurements are cheap and around 20 euros per sample.  
 25 Alternatively, an UV absorbance sensor measuring the UV absorbance at wavelength of 254 or 280 nm can be  
 26 used as a generic sensor providing insights in the reactivity of ozone with the organic matter (Westerhoff et al.,  
 27 1999). Besides direct measurement or a generic sensor, Langergraber et al. (2003) developed a soft-sensor allowing  
 28 to estimate the DOC concentration based on measured UV/VIS wavelengths and by applying principal component  
 29 analysis followed by partial least squares regression. These soft-sensors do require to be calibrated locally based  
 30 on an obtained dataset from lab measurements. The UV/Vis sensor is, besides regular cleaning, easy to maintain,  
 31 and less than half the price of a specific TOC sensor. Besides DOC, UV<sub>254</sub> also determines the efficiency of ozone  
 32 and should therefore be monitored continuously. A specific on-line sensor is available which only measures  
 33 UV<sub>254</sub>, is easy to use, sensitive and low in maintenance and costs. An alternative generic sensor is the UV/Vis  
 34 sensor which measures all wavelengths between 200-735 nm. This should only be used instead if the sensor is  
 35 used to measure other parameters, such as DOC, as well.  
 36



1 **AOC, bromate and bromide**

2 AOC and bromate are disinfection by-products formed during ozonation. Depending on the influent concentrations  
3 of DOC and bromide and the amount of ozone dosed, the AOC and bromate concentration are determined. There  
4 is no on-line sensor available for measuring the AOC concentration in accordance with the approved standard  
5 methods (Eaton et al., 2005). AOC is one of the disinfection by-products that needs to be monitored. A change in  
6 organic matter composition and/or ozone dose will directly result in a change in AOC concentration, therefore  
7 requiring on-line monitoring in the effluent of the ozone step. AOC is subsequently biodegraded in BAC filtration  
8 step and enhances the microbiological activity in the filters. Ross et al. (2019) showed that a sudden change in  
9 AOC concentration does not result in a direct deterioration of the effluent quality of the BAC filters. Therefore, a  
10 continuous monitoring of the AOC concentration in the effluent of the BAC filter is not required. The lab  
11 measurements are high in costs, due to the labour intensity of the analysis. Hammes and Egli (2005) developed a  
12 quicker laboratory method to determine the AOC concentration using flow cytometry. Until now this method is  
13 only available as off-line method and therefore not suitable for on-line monitoring. The water quality model  
14 developed by van der Helm et al. (2009) is able to predict the formation of disinfection by-products such as AOC  
15 by using Matlab/Simulink®. Another soft-sensor is the software algorithm published by Ross et al. (2016) that  
16 uses different UV/Vis wavelengths to predict the AOC formation.  
17

18 There are no on-line sensors available for measuring the bromate and bromide concentration. Bromate needs to be  
19 monitored for compliance since it is possibly carcinogenic and is not removed in existing downstream treatment  
20 steps. A change in bromide concentration or a change ozone dose can impact the bromate concentration directly.  
21 The bromide levels in the influent of the Weesperkarspel treatment plant have been very stable, requiring no need  
22 for continuous monitoring. Since the bromate levels can change with changing ozone dose, on-line monitoring of  
23 bromate in the effluent of the ozone step is proposed. The lab measurements are moderate in costs, due to the  
24 reagents required. Van der Helm et. al. (2008a) found a linear relationship between the bromate concentration and  
25 Ct value, allowing the Ct value to be a surrogate parameter once the initial bromate concentration is known.  
26 Cromphout et al. (2013), found a linear relationship between ozone dose, temperature and bromate formation.  
27 These models can be used to predict the bromate concentration based on the ozone dosed, temperature, pH and  
28 bromide concentration in the influent. Another available soft-sensor is the software algorithm published by Ross  
29 et al. (2016) using different UV/Vis wavelengths to determine the Ct value and bromate formation. It should be  
30 tested till what extent these algorithms can be locally calibrated for changing bromide concentrations.  
31

32 **Pathogenic micro-organisms and ozone concentration in water**

33 There are no on-line sensors available to specifically measure a certain pathogenic microorganism. The lab  
34 measurements are high in costs, due to labour intensity of the analysis. The pathogenic microorganism  
35 concentration in the influent together with above parameters do determine the required ozone dosage and therefore  
36 require continuous monitoring. The USEPA (1989) published Ct values for determining the log inactivation of  
37 pathogenic microorganisms for different water temperatures. This allows the Ct value to be used as a surrogate  
38 parameter if the influent concentration is known. The water quality model developed by van der Helm et al. (2009)  
39 is able to predict the Ct value based on above measured parameters and applied ozone dose. In addition, Ross et  
40 al. (2016) published a software algorithm that uses different UV/Vis wavelengths to determine the Ct value.  
41 Verification via lab analysis of pathogenic microorganisms on a weekly/monthly basis, depending on the  
42 variability of the source water quality, will help determine the log inactivation and associated Ct value to be  
43 achieved. Besides using soft-sensors to determine the Ct value based on a change in UV/Vis pattern, the ozone in  
44 water can be determined by on-line measurements. These measurements do require local calibration by means of  
45 lab measurements. It is an easy and sensitive measurements that does require regular maintenance to prevent  
46 biofouling. Cost of on-line and lab measurements are moderate due to the calibration fluid required. In order to be  
47 able to determine the Ct value based on the ozone in water concentrations, multiple sampling points are required  
48 in space.  
49

50 **Phosphate and nitrogen**



1 Phosphate, nitrogen and carbon are the nutrients required for the microbiology in the BAC filters to grown on.  
2 Phosphate is a frequently on-line measured and controlled parameter in wastewater environments. The available  
3 on-line measurements are easy to use, sensitive enough, but do require regular maintenance due to reaction agents  
4 used. The costs of both lab and on-line application are moderate. To the authors knowledge there are no on-line  
5 nitrogen measurements available. The costs of lab measurements are moderate. In the current treatment plant setup  
6 there is no option to alter the phosphate or nitrogen concentration (by means of dosing) and as a result there is no  
7 need to continuously monitor these concentrations in the influent of the BAC filters.

### 8 9 **Viability bacterial cells**

10 Viable bacterial cells are present in the surface water. During ozonation typically disinfection of viable bacterial  
11 cells takes place, which subsequently can regrow in following treatment steps (Vital et al., 2012). The  
12 determination of viable bacterial cells has developed in the last couple of years from a laborious intensive  
13 measurement using microscopy, to rapid determination in the lab using flow cytometry to customizing the flow  
14 cytometry equipment for on-line applications (Besmer et al., 2014; Besmer et al., 2017). Ross et al. (2019) showed  
15 that the effect of viable bacterial cells in the influent of the BAC filters is limited in respect to the performance of  
16 the BAC filters, therefore discarding the need for on-line monitoring. The costs of both lab and on-line  
17 measurements are still high but expected to reduce in future as per the innovation taking place to enhance rapid  
18 detection.

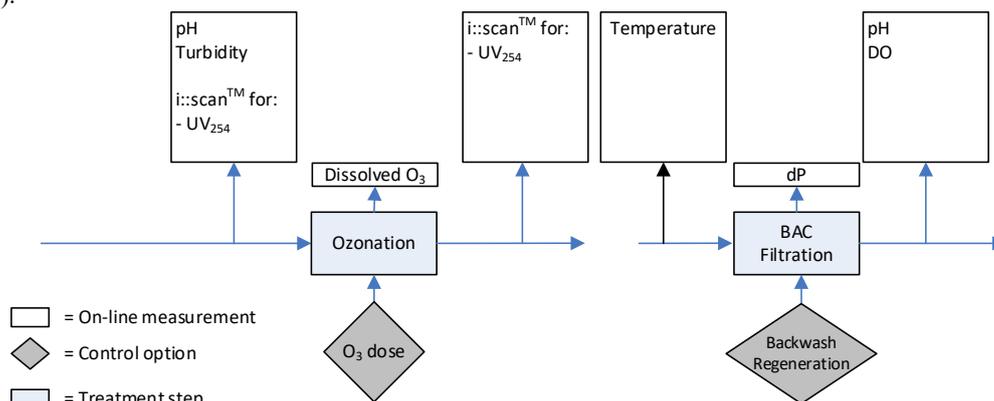
### 19 20 **Pressure drop**

21 The pressure drop is typically measured to determine the clogging ratio in the filter bed. Pressure drop  
22 measurements are available on-line and have been fully developed. It is an easy measurement, which is sensitive  
23 and low in maintenance. The costs are low. For BAC filtration it is, besides turbidity, the main indicator if a filter  
24 is clogging and needs backwashing. On-line monitoring is therefore required and frequently applied.

### 25 26 **Determination of integrated monitoring strategy of treatment plant**

27 When evaluating the ozonation and BAC filtration step as an integrated system, it is not required to monitor the  
28 AOC in the effluent of the ozonation due to the robustness of the BAC filtration step (Ross et al., 2019). The DO  
29 concentration in the influent of the BAC filter will always be sufficient as a result of the preceding ozonation step,  
30 therefore there is no need to continuously monitor this concentration in the influent. For Weesperkaspel, the  
31 temperature of the water and pH will not change due to application of ozonation, hence there is no need to monitor  
32 this in the influent of the BAC filters.

33  
34 In Figure 3 the current monitoring strategy of Weesperkaspel is shown. This strategy was adjusted per the  
35 outcomes of the different research described in this paper (van der Helm, 2007; Ross et al., 2016; van Schagen,  
36 2009).



37





1 The main objective of this paper was to develop a design methodology able to determine an optimised water  
2 quality monitoring strategy to support future direct control of the drinking water treatment plant based on incoming  
3 water quality. A seven step approach was defined, and each step was demonstrated for the treatment processes  
4 ozone and BAC filtration. It was shown how the previous on-line water quality monitoring program of the  
5 treatment plant Weesperkarspel was optimised and subsequently can be finetuned in future.

6  
7 Evaluation of available on-line sensors showed that the parameters typically measured to show compliance with  
8 the WHO standards are commonly available. Direct measurements of the more complex parameters such as AOC  
9 and bromate are not available on-line. The use of soft-sensors, able to estimate the bromate and AOC formation,  
10 help to gain continuous on-line data. Besides using soft-sensors as surrogate sensors for parameters currently not  
11 available on-line, they can also provide a cost effective alternative when used to determine multiple parameters  
12 required through one single instrument. Examples in this case were the use of UV-Vis sensors for the determination  
13 of UV254 concentration in the influent, the estimation of DOC in influent and effluent, formation of bromate and  
14 AOC during ozonation and estimation of Ct value in the effluent of the ozonation step. The on-line data obtained  
15 by the (soft-) sensors will help the operator to control the treatment plant based on its objectives and provide  
16 continuous information whether the processes are operating within the required operational window.

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#### 20 21 **REFERENCES**

- 22 Adu-Manu, K. S., Tapparello, C., Heinzelman, W., Katsriku, F. A., and Abdulai, J. D.: Water quality monitoring  
23 using wireless sensor networks: Current trends and future research directions, *ACM Transactions on Sensor*  
24 *Networks*, 13, 10.1145/3005719, 2017.
- 25 Banna, M. H., Imran, S., Francisque, A., Najjaran, H., Sadiq, R., Rodriguez, M., and Hoorfar, M.: Online drinking  
26 water quality monitoring: Review on available and emerging technologies, *Critical Reviews in Environmental*  
27 *Science and Technology*, 44, 1370-1421, 10.1080/10643389.2013.781936, 2014.
- 28 Bertelkamp, C., Reungoat, J., Cornelissen, E. R., Singhal, N., Reynisson, J., Cabo, A. J., van der Hoek, J. P., and  
29 Verliefd, A. R. D.: Sorption and biodegradation of organic micropollutants during river bank filtration: A  
30 laboratory column study, *Water Research*, 52, 231-241, <https://doi.org/10.1016/j.watres.2013.10.068>, 2014.
- 31 Besmer, M. D., Weissbrodt, D. G., Kratochvil, B. E., Sigrist, J. A., Weyland, M. S., and Hammes, F.: The  
32 feasibility of automated online flow cytometry for in-situ monitoring of microbial dynamics in aquatic ecosystems,  
33 *Frontiers in Microbiology*, 5, 10.3389/fmicb.2014.00265, 2014.
- 34 Besmer, M. D., Sigrist, J. A., Props, R., Buyschaert, B., Mao, G., Boon, N., and Hammes, F.: Laboratory-scale  
35 simulation and real-time tracking of a microbial contamination event and subsequent shock-chlorination in  
36 drinking water, *Frontiers in Microbiology*, 8, 10.3389/fmicb.2017.01900, 2017.
- 37 Bosklopper, T. G. J., Rietveld, L. C., Babuska, R., Smaal, B., and Timmer, J.: Integrated operation of drinking  
38 water treatment plant at Amsterdam water supply, *Water Science and Technology: Water Supply*, 4, 263-270,  
39 2004.
- 40 Cromphout, J., Goethals, S., and Verdickt, L.: Optimization of the ozone dosage at the drinking water treatment  
41 plant of Kluizen, *Water Science and Technology: Water Supply*, 13, 1569-1575, 10.2166/ws.2013.170, 2013.
- 42 Edzwald, J. K., and Tobiason, J. E.: Enhanced coagulation: US requirements and a broader view, *Water Science*  
43 *and Technology*, 40, 63-70, 1999.
- 44 Elovitz, M. S., Von Gunten, U., and Kaiser, H. P.: Hydroxyl radical/ozone ratios during ozonation processes. II.  
45 The effect of temperature, pH, alkalinity, and DOM properties, *Ozone: Science and Engineering*, 22, 123-150,  
46 2000.
- 47 Escobar, I. C., Randall, A. A., and Taylor, J. S.: Bacterial growth in distribution systems: Effect of assimilable  
48 organic carbon and biodegradable dissolved organic carbon, *Environmental Science and Technology*, 35, 3442-  
49 3447, 2001.
- 50 Graveland, A.: Application of biological activated carbon filtration at Amsterdam water supply, *Water Supply*, 14,  
51 233-241, 1996.



- 1 Gunatilaka, A., and Dreher, J.: Use of real-time data in environmental monitoring: Current practices, *Water*  
2 *Science and Technology*, 47, 53-61, 2003.  
3 <https://www.hach.com/phosphate-analyzers/family?productCategoryId=35546907028>, 2018.  
4 Hammes, F. A., and Egli, T.: New method for assimilable organic carbon determination using flow-cytometric  
5 enumeration and a natural microbial consortium as inoculum, *Environmental Science and Technology*, 39, 3289-  
6 3294, 2005.  
7 IARC: Some naturally occurring and synthetic food components, furocoumarins and ultraviolet relation, IARC  
8 Monographs on the Evaluation of Carcinogenic Risks to Humans, International Agency for Research on Cancer,  
9 Lyon, 1999.  
10 Ikonen, J., Pitkänen, T., Kosse, P., Ciszek, R., Kolehmainen, M., and Miettinen, I. T.: On-line detection of  
11 *Escherichia coli* intrusion in a pilot-scale drinking water distribution system, *Journal of Environmental*  
12 *Management*, 198, 384-392, 10.1016/j.jenvman.2017.04.090, 2017.  
13 Jansen, H. W., Vroegindeweij, A., and Haijma, S.: The role of SCADA systems within integrated process control  
14 systems, *Water Supply*, 15, 43-53, 1997.  
15 Juntunen, P., Liukkonen, M., Lehtola, M. J., and Hiltunen, Y.: Dynamic soft sensors for detecting factors affecting  
16 turbidity in drinking water, *Journal of Hydroinformatics*, 15, 416-426, 10.2166/hydro.2012.052, 2013.  
17 Kano, M., and Nakagawa, Y.: Data-based process monitoring, process control, and quality improvement: Recent  
18 developments and applications in steel industry, *Computers and Chemical Engineering*, 32, 12-24, 2008.  
19 Langergraber, G., Fleischmann, N., and Hofstädter, F.: A multivariate calibration procedure for UV/VIS  
20 spectrometric quantification of organic matter and nitrate in wastewater, *Water Science and Technology*, 47, 63-  
21 71, 2003.  
22 Lazarova, V., and Manem, J.: Biofilm characterization and activity analysis in water and wastewater treatment,  
23 *Water Research*, 29, 2227-2245, Doi: 10.1016/0043-1354(95)00054-o, 1995.  
24 Molina, L. T., and Molina, M. J.: Absolute absorption cross sections of ozone in the 185-350 nm wavelength  
25 region, *Journal of Geophysical Research*, 91, 14,501 - 514,508, 1986.  
26 Nopens, I., Benedetti, L., Jeppsson, U., Pons, M. N., Alex, J., Copp, J. B., Gernaey, K. V., Rosen, C., Steyer, J. P.,  
27 and Vanrolleghem, P. A.: Benchmark Simulation Model No 2: Finalisation of plant layout and default control  
28 strategy, *Water Science and Technology*, 62, 1967-1974, 2010.  
29 Poch, M., Comas, J., Rodríguez-Roda, I., Sánchez-Marré, M., and Cortés, U.: Designing and building real  
30 environmental decision support systems, *Environmental Modelling & Software*, 19, 857-873,  
31 <https://doi.org/10.1016/j.envsoft.2003.03.007>, 2004.  
32 Roccaro, P., Chang, H.-S., Vagliasindi, F. G. A., and Korshin, G. V.: Differential absorbance study of effects of  
33 temperature on chlorine consumption and formation of disinfection by-products in chlorinated water, *Water*  
34 *Research*, 42, 1879-1888, 10.1016/j.watres.2007.11.013, 2008.  
35 Ross, P. S., van der Helm, A. W. C., van den Broeke, J., and Rietveld, L. C.: On-line monitoring of ozonation  
36 through estimation of Ct value, bromate and AOC formation with UV/Vis spectrometry, *Analytical Methods*, 8,  
37 3148-3155, 10.1039/c5ay03308j, 2016.  
38 Ross, P. S., van der Aa, L. T. J., van Dijk, T., and Rietveld, L. C.: Effects of water quality changes on performance  
39 of biological activated carbon (BAC) filtration, *Separation and Purification Technology*, 212, 676-683,  
40 <https://doi.org/10.1016/j.seppur.2018.11.072>, 2019.  
41 Schlegel, S., and Baumann, P.: Requirements with respect to on-line analyzers for N and P, *Water Science and*  
42 *Technology*, 33, 139-146, 10.1016/0273-1223(96)00166-7, 1996.  
43 Sereďyńska-Sobecka, B., Tomaszewska, M., Janus, M., and Morawski, A. W.: Biological activation of carbon  
44 filters, *Water Research*, 40, 355-363, 2006.  
45 Servais, P., Billen, G., and Bouillot, P.: Biological colonization of granular activated carbon filters in drinking-  
46 water treatment, *Journal of Environmental Engineering*, 120, 888-899, 1994.  
47 Simpson, D. R.: Biofilm processes in biologically active carbon water purification, *Water Research*, 42, 2839-  
48 2848, 2008.  
49 Sontheimer, H., Crittenden, J. C., and Summers, R. S.: Activated carbon for water treatment, 2nd ed., AWWA -  
50 DVGW Forschungsstelle Engler Bunte Institut, Karlsruhe, Germany, 1988.  
51 Bromate Analysis Methods, 2018.



- 1 USEPA: Guidance manual for compliance with the filtration and disinfection requirements for public water
- 2 systems using surface water supplies, Washington D.C., 1989.
- 3 USEPA: Edition of the drinking water standards and health advisories table, EPA-822-F-18-001, Office of Water
- 4 U.S. Environmental Protection Agency, Washington, DC, 2018.
- 5 van de Ven, W., Bakker, S., Wuestman, R., McEwan, M., Mazier, S., Bergmans, B., Ross, P., Rietveld, L., and
- 6 van Schagen, K.: Quality control for groundwater treatment plant Oldeholtspade: strategies for modeling and
- 7 process management, IWA World Water Congress and Exhibition, Montreal, Canada, 2010,
- 8 van den Broeke, J., Ross, P. S., van der Helm, A. W. C., Baars, E. T., and Rietveld, L. C.: Use of on-line UV/Vis-
- 9 spectrometry in the measurement of dissolved ozone and AOC concentrations in drinking water treatment, *Water*
- 10 *Science and Technology*, 57, 1169-1175, 2008.
- 11 van der Aa, L. T. J., Rietveld, L. C., and van Dijk, J. C.: Effects of ozonation and temperature on the biodegradation
- 12 of natural organic matter in biological granular activated carbon filters, *Drinking Water Engineering and Science*,
- 13 4, 25-35, 2011.
- 14 van der Helm, A. W. C.: Integrated modeling of ozonation for optimization of drinking water treatment, PhD,
- 15 *Sanitary Engineering*, Delft University of Technology, Delft, 151 pp., 2007.
- 16 van der Helm, A. W. C., Rietveld, L. C., Baars, E. T., Smeets, P. W. M. H., and van Dijk, J. C.: Modeling
- 17 disinfection and by-product formation during the initial and the second phase of natural water ozonation in a pilot-
- 18 scale plug flow reactor, *Journal of Water Supply: Research and Technology - AQUA*, 57, 435-449, 2008a.
- 19 van der Kooij, D., Hijnen, W. A. M., and Kruithof, J. C.: Effects of ozonation, biological filtration and distribution
- 20 on the concentration of easily assimilable organic carbon (AOC) in drinking water, *Ozone: Science and*
- 21 *Engineering*, 11, 297-311, 1989.
- 22 van der Kooij, D.: Assimilable organic carbon as an indicator of bacterial regrowth, *Journal American Water*
- 23 *Works Association*, 84, 57-65, 1992.
- 24 van Schagen, K., Rietveld, L., Veersma, A., and Babuška, R.: Control-design methodology for drinking-water
- 25 treatment processes, *Water Science and Technology: Water Supply*, 10, 121-127, 2010.
- 26 van Schagen, K. M., Rietveld, L. C., and Babuška, R.: Dynamic modelling for optimisation of pellet softening,
- 27 *Journal of Water Supply: Research and Technology - AQUA*, 57, 45-56, 10.2166/aqua.2008.097
- 28 10.1016/j.watres.2007.07.019; Schock, M., Temperature and ionic strength correction to the langelier index-
- 29 revisited (1984) *J. Am. Wat. Wks. Assoc.*, 76, pp. 72-76; Wiechers, H., Sturrock, P., Marais, G., Calcium carbonate
- 30 crystallization kinetics (1975) *Wat. Res.*, 9, pp. 835-845, 2008.
- 31 van Schagen, K. M.: Model-Based Control of Drinking-Water Treatment Plant PhD, Delft University of
- 32 Technology, 2009.
- 33 Vanrolleghem, P. A., and Lee, D. S.: On-line monitoring equipment for wastewater treatment processes: State of
- 34 the art, *Water Science and Technology*, 47, 1-34, 2003.
- 35 Verliefde, A., Cornelissen, E., Amy, G., Van der Bruggen, B., and van Dijk, H.: Priority organic micropollutants
- 36 in water sources in Flanders and the Netherlands and assessment of removal possibilities with nanofiltration,
- 37 *Environmental Pollution*, 146, 281-289, 10.1016/j.envpol.2006.01.051, 2007.
- 38 Vital, M., Dignum, M., Magic-Knezev, A., Ross, P., Rietveld, L., and Hammes, F.: Flow cytometry and adenosine
- 39 tri-phosphate analysis: Alternative possibilities to evaluate major bacteriological changes in drinking water
- 40 treatment and distribution systems, *Water Research*, 46, 4665-4676, 2012.
- 41 von Gunten, U.: Ozonation of drinking water: Part I. Oxidation kinetics and product formation, *Water Research*,
- 42 37, 1443-1467, 2003a.
- 43 von Gunten, U.: Ozonation of drinking water: Part II. Disinfection and by-product formation in presence of
- 44 bromide, iodide or chlorine, *Water Research*, 37, 1469-1487, 10.1016/s0043-1354(02)00458-x, 2003b.
- 45 Westerhoff, P., Aiken, G., Amy, G., and Debroux, J.: Relationships between the structure of natural organic matter
- 46 and its reactivity towards molecular ozone and hydroxyl radicals, *Water Research*, 33, 2265-2276, 1999.
- 47 WHO: Guidelines for Drinking Water Quality, Geneva, 2008.
- 48 Wiersema, Y.: Efficiency and efficacy of ozonation for disinfection at the Weesperkarspel drinking water
- 49 treatment plant, MSc, Utrecht University, Utrecht, 2018.
- 50